

# WJEC (Eduqas) Chemistry A-level

## SP OA2.2 - Identification of Aldehydes/Ketones by their Reaction with 2,4-Dinitrophenylhydrazine

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### Aim

To use **2,4-dinitrophenylhydrazine** (2,4-DNPH) to identify **aldehydes** and **ketones**.

### Apparatus and Chemicals

- 5 x Pasteur pipette
- 2 x spatula
- 4 x boiling tube
- Test-tube rack
- 100 cm<sup>3</sup> beaker
- 250 cm<sup>3</sup> beaker
- 10 cm<sup>3</sup> measuring cylinder
- 100 cm<sup>3</sup> conical flask
- Hot water bath
- Buchner funnel
- Suction apparatus
- Filter paper
- Melting point apparatus (or Thiele tube)
- Capillary tube
- 2,4-DNPH
- 1.0 mol dm<sup>-3</sup> H<sub>2</sub>SO<sub>4</sub>
- Butanone (CH<sub>3</sub>CH<sub>2</sub>COCH<sub>3</sub>)
- Benzaldehyde (C<sub>6</sub>H<sub>5</sub>CHO)
- CH<sub>3</sub>CH<sub>2</sub>OH

### Safety Considerations

- ★ 2,4-DNPH - flammable, toxic
- ★ CH<sub>3</sub>CH<sub>2</sub>COCH<sub>3</sub> - flammable, irritant
- ★ C<sub>6</sub>H<sub>5</sub>CHO - harmful
- ★ 1.0 mol dm<sup>-3</sup> H<sub>2</sub>SO<sub>4</sub> - irritant
- ★ CH<sub>3</sub>CH<sub>2</sub>OH - highly flammable





## Method

1. Add about **1 cm depth** of  $\text{CH}_3\text{CH}_2\text{COCH}_3$  to a **boiling tube**.
2. Using a measuring cylinder, carefully add  $8 \text{ cm}^3$  of **2,4-DNPH** solution to the boiling tube.
3. If a precipitate does not form, carefully add dilute sulfuric acid dropwise until a **precipitate is formed**.
4. Filter the mixture under **reduced pressure** to isolate the solid derivative.
5. Purify the derivative by **recrystallisation** from the minimum amount of hot ethanol.
6. Filter the purified derivative under **reduced pressure**.
7. **Dry** a small sample of the derivative by pressing between filter papers.
8. Obtain a **melting point** for this sample.
9. Repeat steps 1-8 with  $\text{C}_6\text{H}_5\text{CHO}$ .

